CHEM F321 Organic Chemistry I

Fall 2019

CRN(s): 73922, 73923, 73924, 73925, 73926, 73927

Credits: 4 credits

Lecture: REIC 201, MWF 8:00-9:00 AM (in person)

Lab: REIC 241, Various Times

Prerequisite: Chem F106X with grade C or better

Lecture Instructor



Jennifer Guerard, Ph.D. Office: REIC 180 Office Hrs: W 9-11 am Phone: (907) 474-5231 Email: jguerard@alaska.edu

Lab Instructor



Thomas Green, Ph.D. Office: REIC 174 Office Hrs: T, W 1-2pm Phone: (907) 474-1559 Email: tkgreen@alaska.edu

Lab syllabus TBD

Catalog Course Description A systematic study of the more important functional groups of carbon compounds, including their mechanisms of reaction, methods of synthesis, and physical and spectroscopic properties. Lab portion will include an introduction to synthetic techniques and spectroscopy.



Course Goals The goals of this course are to understand the fundamental concepts of bonding of organic compounds, how conformations of hydrocarbons relate to stability, basic concepts of stereochemistry. reactions and associated mechanisms of hydrocarbons, and the use of spectroscopic techniques to determine structure of organic molecules.

Course Structure Coursework will follow topics in the order described on the Tentative Lecture Schedule. The instructor will lecture using slides and/or whiteboard, providing copies of notes to the students via Blackboard. Clickers, quizzes, homework, in-class activities, and exams will assess student understanding of concepts. Lab schedule/syllabus will be handed out during lab sections.

Course Materials

Required Materials:

Lecture Text: Karty, J. Organic Chemistry: Principles and Mechanisms 2nd ed. Norton, 2018 Print versions come with solutions

manual study guide - super helpful Laboratory text: Padias, A. Making the Connections: A How-to Guide for Organic

Chemistry Lab Techniques, 3rd ed.

Other Required Materials:

- Lab book (see details in Blackboard)
- NON-graphing scientific calculator
- TopHat Subscription
- Alchemie Mechanisms
- o Internet/computer access (available in REIC 172), regularly checking Blackboard and email

HIGHLY Recommended Materials:

- Klein, D. Organic Chemistry as a Second Language: First Semester Topics, 4th ed. Wiley, 2016
- Eubanks, I. Dwaine. Preparing for Your ACS Examination in Organic Chemistry: The Official
- Molecular model kit my favorite is in the UAF bookstore











Important Dates:

Aug 26 First day of class
Sep 7 Last day for late
registration, drops
with no
appearance on
academic record

Sep 20 EXAM I Oct 14 EXAM II

Nov 1 Last day for withdrawal with W

Nov 8 EXAM III

Nov 25 EXAM IV

Dec 6 Last day of class

Dec 9 FINAL 8-10am

Dec 18 Grades Posted

Student Learning Outcomes A more detailed list of more specific learning outcomes for each chapter is posted on Blackboard and sent via Blackboard before each quiz. Broadly, at the end of this course, students should be to:

- Identify and draw common functional groups
- Understand fundamental concepts of bonding and acidity
- Name a variety of organic functions groups, including hydrocarbons.
- Predict the reactivity of organic compounds involving nucleophilic and electrophilic substitution, elimination, and addition mechanisms.
- Know common reagents used for hydrocarbon transformation into other functional groups.
- Interpret IR and NMR spectra of organic compounds to arrive at a structure.
- Draw and interpret 3D structures of stereoisomers.
- Predict and write out mechanisms of reactions based on fundamental concepts of acid/base chemistry (nucleophiles/electrophiles).

Evaluation and Grade Assignment:

A = 900 - 1000 points B = 800 - 899 points

C = 700 - 799 points

D = 600 - 699 points

F = 0 - 599 points

Final Grade Rounding Policy: If cumulative final grade is within 5 points of a grade point cutoff **AND** final exam percentage falls into a letter grade above that from the cumulative letter grade, then final grade will be rounded up to next letter grade.

Your instructor follows the UAF Incomplete Grade Policy:

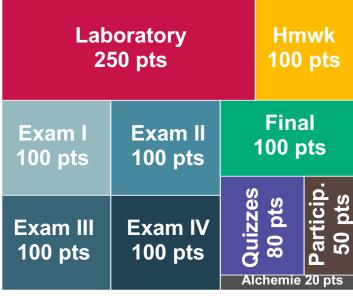
Your instructor follows the UAF Incomplete Grade Policy:

"Incomplete" is a temporary grade used to indicate that the student has satisfactorily completed (C or better) the majority of work in a course but for personal reasons beyond the student's control, such as sickness, has not been able to complete the course during the regular semester. Negligence or indifference are not acceptable reasons for an "IN" grade."

Notes and Policies

Homework (Hmwk) consists of near-daily assignments derived from the textbook and graded for completion, each out of 10 points. Expect homework to take *at least* a few hours each week. All listed assignments are posted on Blackboard and are due at the start of class (either turned in in class, emailed, or to professor's mail box in REIC 194 – make sure any scans, photos, or written assignments are legible and clear to read!). HW will be averaged and scaled to 100 points. Extra credit assignments will average into the total homework score.

Assignments should be written on paper with your name as it appears on UAOnline and Assignment listed at top left. If multiple pages, please staple together and write name on all pages. Label each question clearly and write neatly and legibly. Note some questions have been abbreviated in homework compared to textbook. Only listed question parts in homework packet are required.



Hmwk Extension Policy: A 2-day extension period is allowed to apply either all to one homework of a student's choice or to split into two 1-day extensions on two assignments of their choice. In these cases, this must be communicated to the professor before assignment due date by email, and then assignment must be turned in via methods above by 8:00 am on the extended due date.

Exams allow the use of molecular model kits and non-programmable calculators only. You must turn in your exam before leaving the room. The final exam will be the comprehensive multiple-choice exam, covering the entire semester content.

Quizzes. 9 quizzes (lowest score dropped) will be administered by paper or TopHat, each worth 10 points for a total of 80 points (see syllabus schedule; 5 questions each, ~15 min). Students will be notified ahead of time of the format for upcoming quizzes.

Class Participation. Several in-class activities involve individual responses using TopHat on an internet connected or text-capable device. It is thus important to attend and bring your device to class every day. It is the student's responsibility to bring the device to each class, replace if lost, verify it is registered correctly on the instructor's database, and keep in functioning condition. It is ultimately the student's responsibility to address problems with their device and/or check with the instructor concerning their clicks. TopHat registration must occur by SUNDAY, Sept 8th, 11:59 PM. Registration directions are posted on Blackboard. If you do not own a compatible device please notify the instructor for appropriate tech. accommodations. Participation is graded out of 80% of possible opportunities starting class Sept 9th.

Alchemie Mechanisms. Instructions are on Blackboard under Course Materials for App download / registration. If you do not own a compatible device please notify the instructor for appropriate tech. accommodations.

Instructor-Initiated Withdrawals. Until Friday, Nov 1st, the instructor has the right to withdraw a student who has not participated substantially in the course. Any (or combination) of the following constitute non-participation: 1) Missed exam or 2 quizzes without excused absence, 2) At least 2 incomplete homework assignments, or 3) Less than 50% attendance as registered by TopHat participation.

Mobile Devices and Laptops. Mobile devices are used during class for participation, worksheet access if desired, and course activities. Devices must be turned to

Support & Accommodations

Chemistry Learning Center (CLC) – CHEM 321 TAs keep regular office hours, which can be found on the CLC calendar: www.uaf.edu/chem/clc/ and will be posted on Blackboard once available. Chemistry computer lab (REIC 172) is available for course related activities – www.uaf.edu/chem/instrumentation/policies

Student Protections and Services Statement. Every qualified student is welcome in my classroom. As needed, I am happy to work with you, disability services, veterans' services, rural student services, etc. to find reasonable accommodations. Students at this university are protected against sexual harassment and discrimination (Title IX), and minors have additional protections. As required, if I notice or am informed of certain types of misconduct, then I am required to report it to the appropriate authorities. For more information on your rights as a student and the resources available to you to resolve problems, please go the following site: www.uaf.edu/handbook/.

Veteran Support Services. Walter Crary (wecrary@alaska.edu) is the Veterans Service Officer at the Veterans Resource Center, 111 Eielson Building, 474-2475. Fairbanks Vet Center: 456-4238. VA Community Based Outpatient Clinic at Ft. Wainwright: 361-6370.

silent or "vibrate" mode during class. Only usage of electronic devices that facilitates learning is permitted.

Absences. Make-up exams/quizzes are only allowed in the event of a legitimate cause and proper notification to the instructor. Known absences (e.g., intercollegiate sports, travel for military or university purposes) are required to provide notification by email before the absence; unexpected/emergency absences (e.g., illness, family, or personal emergency) need to provide notification at the earliest possible opportunity. Students may be asked to provide documentation for the absence. Make-ups should be scheduled as soon as possible from the originally scheduled date. Summary: If you have any absences (scheduled or unscheduled), or questions about absences/make-ups, email your instructors: Dr. Guerard (jguerard@alaska.edu) for lecture, and Dr. Green (tkgreen@alaska.edu) for lab.

Ethical Considerations and Honor Code. Students must adhere to UAF policies, the student code of conduct as well as the University of Alaska Honor Code. The Chemistry and Biochemistry Department Policy on Cheating states: Any student caught cheating will be assigned a course grade of F. The student's academic advisor will be notified of this failing grade and the student will not be allowed to drop the course. Use of cell phones or electronic devices other than a non-programmable scientific calculator during exams constitutes cheating and will result in an F in the course.

UA is an AA/EO employer and educational institution and prohibits illegal discrimination against any individual: <u>alaska.edu/nondiscrimination</u>.

Disabilities Services. The Office of Disability Services implements the Americans with Disabilities Act (ADA), and ensures that UAF students have equal access to the campus and course materials. I will work with the Office of Disabilities Services (*208 Whitaker, 474-5655) to provide reasonable accommodation to students with disabilities.

Student Support Services. The Student Support Services (SSS) program (trio.sss@alaska.edu), located in 512 Greuning (474-6844), provides opportunities for academic development, assists students with college requirements, and serves to motivate students toward successful completion of their degree programs. For more information: 474-6844 www.uaf.edu/sss/

Amending this Syllabus: Before the drop date, I may revise the syllabus to correct for any errors. Revision at a later time would require majority vote by students present in class on day issue is decided. Adjustments to the lecture schedule and homework due dates may be made

at the instructor's discretion anytime during the semester. Any revisions will be distributed via Blackboard and announced in class.



CHEM F321 Fall 2019 Tentative Schedule

| | Wk | Date | Topic | Assigned Readings | Klein | Assignments & Quizzes |
|----|---------------|--------------------|----------------------|--|---------|---|
| 1 | М | 8/26 | Gen. Chem. | 1a: 1.1 – 1.6 | Ch 1,2 | 7.65.95 & Gall=255 |
| • | W | 8/28 | Review | 1b: 1.7-1.11 | 011 1,2 | Alchemie #1 start |
| | F | 8/30 | TICVICW | 1c: 1.12-1.14 | | Alchemie #2 start |
| 2 | W | 9/4 | Gen. Chem. | Ch 2: 2.1-2.11 | | Quiz 1, HW 1ab due |
| _ | F | 9/6 | Review | Ch 3: 3.1-3.11 | Ch 4 | HW 1c due |
| 3 | <u>.</u> М | 9/9 | Nomenclature | IC A, B | Ch 5 | HW 2 due |
| 0 | W | 9/11 | Nomencialure | IC E | On o | Quiz 2, HW 3 due |
| | F | 9/13 | Conformations | 4a: 4.1-4.7 | Ch 6 | HW IC AB due |
| 4 | M | 9/16 | Comornations | 4b: 4.8-4.13 | 011 0 | HW IC E due |
| 7 | W | 9/18 | | Practice Problems | | HW 4a due |
| | • • • | 0/10 | | 1 Tablice 1 TobleTTib | | Alchemie #1, 2 due 11:30 pm 9/19 |
| | F | 9/20 | | EXAM I: Ch 1-4, IC AE | R F | HW 4b due |
| 5 | M | 9/23 | Chirality | 5a: 5.1-5.4, IC C | Ch 7 | 1111 15 440 |
| 3 | W | 9/25 | Officiality | 5b: 5.5-5.7, 5.13-5.15 | OH 7 | |
| | F | 9/27 | | 5c: 5.8-5.12 | | Quiz 3, HW 5a due |
| 6 | <u>.</u> М | 9/30 | Acidity | 6a: 6.1-6.3, 6.10-6.11 | Ch 3 | HW 5b due, Alchemie #3 start |
| O | W | 10/2 | Acidity | 6b: 6.4-6.9 | On o | HW 5c due, Alchemie #4 start |
| | F | 10/2 | Element. Steps | 7a: 7.1-7.4 | Ch 8 | Quiz 4, HW 6a due, Alchemie#5 start |
| 7 | M | 10/7 | Liement: Otopo | 7b: 7.5-7.8 | 0110 | HW 6b due, Alchemie #6 start |
| ' | W | 10/7 | IR | 15: 15.4-15.6 | Ch 7 | HW 7a due |
| | F | 10/3 | 11 (| Practice Problems | OH I | Quiz 5, HW 7b due |
| | Г | 10/11 | | Fractice Froblems | | • |
| _ | | 40/44 | | EVANUE OF E 7 10 0 | | Alchemie #3-6 due 11:30 pm 10/13 |
| 8 | M \A/ | 10/14 10/16 | SN1/E1 | EXAM II: Ch 5-7, IC C 8a: 8.1-8.4 | , ט | Alchemie #7 start |
| | W F | 10/18 | SINI/EI | 8b: 8.5-8.7 | | |
| 9 | | | CN11/CN10/E1/E0 | | Ch 1 | HW 9a due |
| 9 | M W | 10/21 10/23 | SN1/SN2/E1/E2 | 9a: 9.1-9.9 9b: 9.10-9.14 | Ch 4 | HW 8a due |
| | F | 10/25 | | Practice Problems | CII 4 | HW 8b due |
| 10 | | 10/23 | SN1/SN2/E1/E2 | 10a: 10.1-10.4 | Ch 5 | Quiz 6, HW 9a due Alchemie #9 start |
| 10 | M W | 10/28 | SIN 1/SIN2/E 1/E2 | 10a. 10.1-10.4 10b: 10.5-10.7 | CH 5 | |
| | F | 11/1 | | 10c: 10.8-10.10 | | HW 9b due, Alchemie #10 start HW 10a due |
| 11 | M | 11/4 | | Practice Problems | | |
| 11 | W | 11/4 | NMR | 16: 16.1-16.10, 16.13, | 16 16 | Quiz 7, HW 10b due HW 10c due |
| | VV | 11/0 | INIVII | 10. 10.1-10.10, 10.13, | 10.10 | Alchemie #7-10 due 11:30 pm 11/7 |
| | F | 11/8 | | EXAM III: CH 8-10, IR | | Alchemie #1-10 due 11:00 pm 11/1 |
| 12 | M | 11/11 | Electrophilic | 11a: 11.1-11.5 | Ch 10 | Alchemie #11 start |
| 12 | W | 11/13 | Addition I | 11b: 11.6-11.8 | OII IO | HW 16 due |
| | F | 11/15 | Addition | 11c: 11.9-11.12 | | Quiz 8, HW 11a due |
| 13 | M | 11/18 | Electrophilic | 12a: 12.1-12.3 | | HW 11bc due, Alchemie #12 start |
| 13 | W | 11/10 | Addition II | 12b: 12.4-12.5 | | HW 12a due |
| | F | 11/20 | Addition | 120: 12.4-12.5 12c: 12.6-12.8 | | Quiz 9, HW 12b due |
| | 1 | 1 1/44 | | 120. 12.0-12.0 | | Alchemie#11-12 due 11:30pm 11/24 |
| 14 | M | 11/25 | | Evam IV (Ch 11 10 NI | MD) | HW 12c due |
| 15 | | | Organia | Exam IV (Ch 11-12, N 13a: 13.1-13.3 | ivin) | TIW 120 due |
| 15 | M W | 12/2 12/4 | Organic Synthesis | 13a: 13.1-13.3 13b: 13.4-13.8 | | |
| | ۷۷ F | 12/4 | Syriu iesis | Practice Problems | | HW 13ab due |
| | | | COMPDEHENSI | | | |
| | M | 12/9 | CONPREHENSI | VE FINAL 8-10 AM | | HW xc due (rev.) |